

# Fabrication of Nano/Micro Structure on Metal and Semiconductor by Anodization

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## Outline

### ~Old materials with new applications~

We are studying on inorganic surface chemistry and material science, including electrochemistry, surface treatment as well as micro- and nanostructuring. Preparation, characterization and functionalization of nano-objects are important research subject to improve the fundamental understanding of various processes and the resulting materials. The fabrication technique and various materials developed in our laboratory have potential technological and scientific applications in various research fields.

## What is Anodization ?

**Anodization** is an electrochemical process that converts the metal surface into a decorative, durable, corrosion-resistant, anodic oxide finish. Anodizing of aluminum is well-known and widely used, although other nonferrous metals such as magnesium and titanium also can be anodized.

### ■ Anodization of valve metals and functionalization of anodic oxide films

Barrier-type anodic oxide film

- Electrolytic capacitor
- Improvement of dielectric properties

Porous-type anodic oxide film

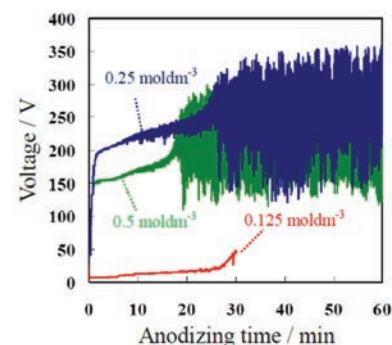
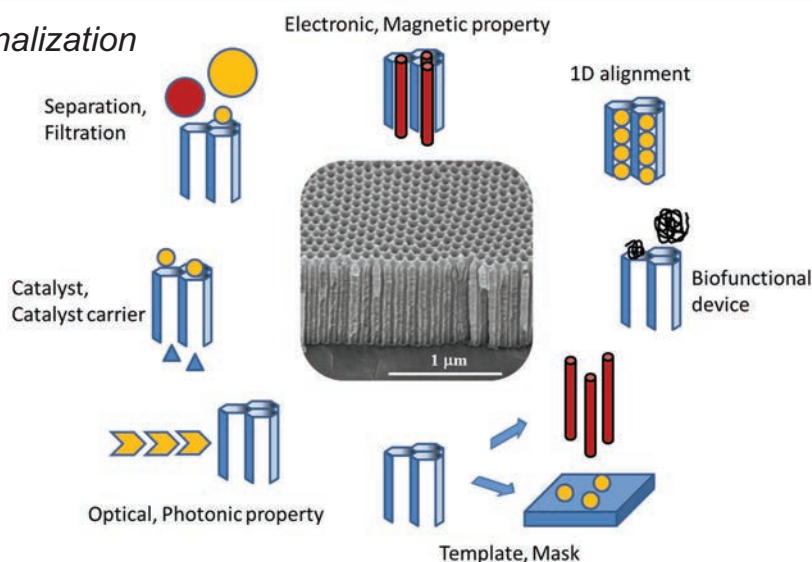
(Al, Ti, Nb, Ta, Sn, Zn, Stainless steel...)

- Protection, decoration of metal surface
- Template, catalyst support, photocatalyst, chemical sensor

### ■ Corrosion protection of magnesium

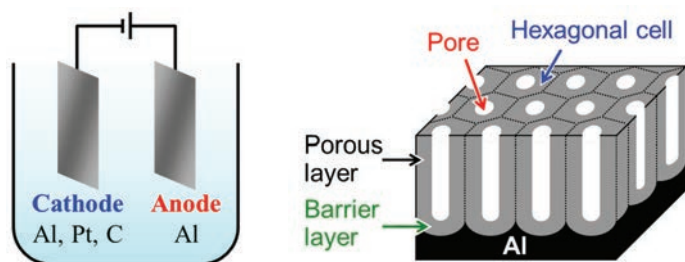
Surface modification, Plasma electrolytic oxidation

- Improvement of corrosion resistance of Mg

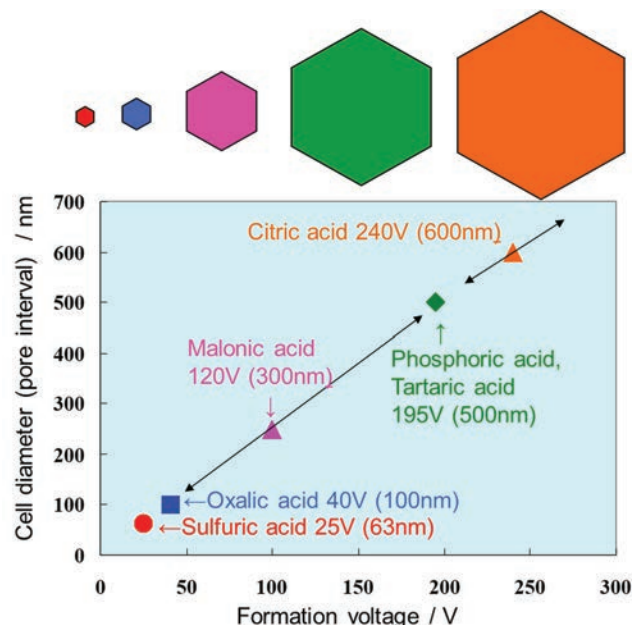


## Characteristics of anodic porous alumina

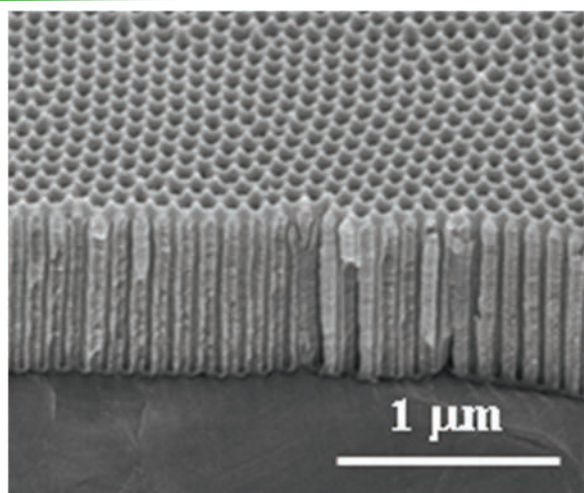
- 1) Controllability of dimensions of porous structure
- 2) Regularity of pore arrangement



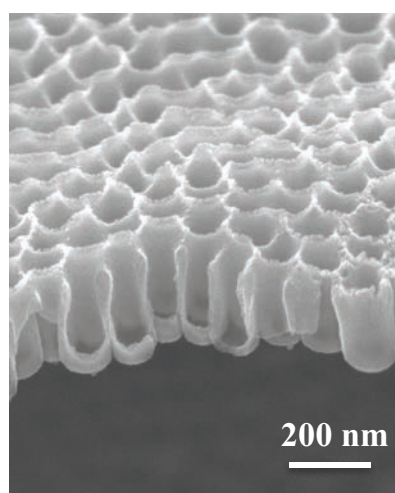
Cell dimension is proportional to formation voltage  
 Cell diameter [nm]  $\approx 2.5$  [nm / V]  $\times$  formation voltage [V]  
 Pore diameter [nm]  $\approx 1.0$  [nm / V]  $\times$  formation voltage [V]



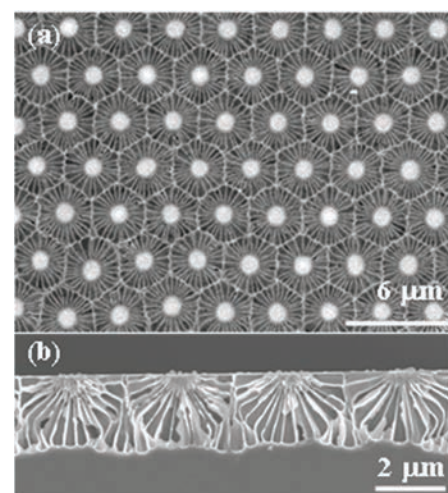
## Application



Porous Al<sub>2</sub>O<sub>3</sub>



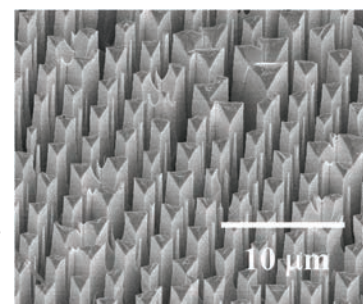
Porous TiO<sub>2</sub>



Porous InP

## Related information

- [Al] S. Ono, M. Saito, M. Ishiguro and H. Asoh  
 Controlling Factor of Self-Ordering of Anodic Porous Alumina  
*Journal of the Electrochemical Society*, **151**, B473 (2004)  
 S. Ono, M. Saito and H. Asoh  
 Self-Ordering of Anodic Porous Alumina Formed in Organic Acid Electrolytes  
*Electrochimica Acta*, **51**, 827 (2005)  
 H. Asoh, M. Ishino and H. Hashimoto  
 Indirect oxidation of aluminum under an AC electric field  
*RSC Advances*, **6**, 90318 (2016)
- [Mg] S. Ono, S. Moronuki, Y. Mori, A. Koshi, J. Liao and H. Asoh  
 Effect of Electrolyte Concentration on the Structure and Corrosion Resistance of Anodic Films Formed on Magnesium through Plasma Electrolytic Oxidation, *Electrochimica Acta*, **240**, 415 (2017)
- [GaAs] H. Asoh, S. Kotaka and S. Ono  
 High-Aspect-Ratio GaAs Pores and Pillars with Triangular Cross Section  
*Electrochemistry Communications*, **13**, 458 (2011)
- [InP] H. Asoh, J. Iwata and S. Ono  
 Hexagonal Geometric Patterns Formed by Radial Pore Growth of InP Based on Voronoi Tessellation  
*Nanotechnology*, **23**, 215304 (2012)



GaAs pillar